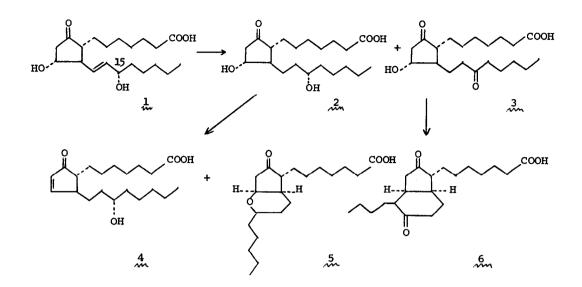
BY-PRODUCTS FORMED IN TRANSFORMATIONS OF PROSTAGLANDIN E₁ R. D. Hoffsommer, D. Taub and N. L. Wendler Merck Sharp & Dohme Research Laboratories Merck & Co., Inc., Rahway, N.J. (Received in USA 30 July 1971; received in UK for publication 28 September 1971)

Hydrogenation of prostaglandin E_1 over platinum group catalysts yields 13,14-dihydroprostaglandin E_1 in only moderate (~45-60%) yield¹. With 10% palladium on carbon in ethyl acetate we obtained 2 in 55-60% yield as well as a more mobile by-product (15-20%) shown to be 9,15-dioxo-lla-hydroxyprostanoic acid 3. Identification of the latter was based on physical measurements, especially the mass spectrum of its methyl ester, which was essentially identical with the published spectrum of 3^2 . Formation of 3 presumably involves double bond migration (Δ -13 = Δ -14) on the catalyst surface at a rate competitive with hydrogenation. This side reaction was avoided by carrying out the hydrogenation over Raney nickel in dioxane, under which conditions 2 was the sole product.

Dehydration of 2 to 13,14-dihydroprostaglandin A₁, 4 in 90% aqueous acetic acid at 60°³ gave, in addition to 4 (45 - 50%) and its 15-acetate (~5%), a more mobile substance (30 - 35%) identified as 11,15-epoxy-9-oxoprostanoic acid 5. Its uv spectrum showed the absence of cyclopentenone functionality. Carboxyl and cyclopentanone functionalities were shown to be present by its ir spectrum, and its nmr spectrum (CDCl₃) indicated the presence of 2H alpha to ether oxygen (1H multiplets at 3.73 and 4.34 b). The mass spectrum of the methyl ester had major peaks at 352 (M⁺), 334 (M⁺-H₂O), 321 (M⁺-OCH₃), 309 (M⁺-C₃H₇), 303 (M⁺-H₂O-OCH₃) and 281 (M⁺-C₅H₁₁). Cyclic ether 5 evidently arises from 4 by intramolecular addition of the C-15 hydroxyl group to the Δ -10 double bond.

The lla-hydroxy-9,15-dione 3 showed no uv absorption after warming in methanolic sodium hydroxide, suggesting formation of the Michael product 6 <u>via</u> the expected intermediate cyclopentenone⁴.

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 Assignment of <u>cis</u> stereochemistry at the ring junctions of 5 and 6 is based on inspection of models.